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A NAPHTHALENIC NORDITERPENE FROM VELLOZACEAE

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Key Word Index—*Vellozia epidendroides*; *V. phalocarpa*; Velloziaceae; naphthalenic norditerpene.

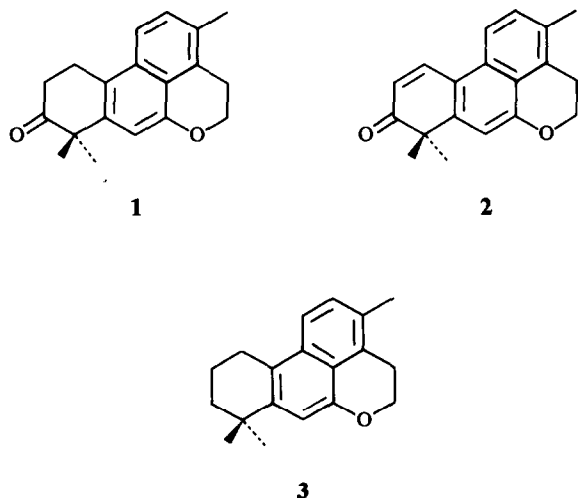
Abstract—The isolation of a new naphthalenic nor-cleistanthane diterpene from *Vellozia epidendroides* and *V. phalocarpa* is described.

Recently we described two naphthalenic diterpenes (1 and 2) from *Vellozia stipitata* L. B. Smith et Ayensu and *V. declinans* Goethart et Henrard [1]. In the present communication we report the isolation of a new naphthalenic nor-cleistanthane diterpene (3) from a hexane extract of roots, stem and leaf-sheaths of *V. epidendroides* Martius ex Schultes and *V. phalocarpa* Pohl.

The molecular formula of 3, $C_{19}H_{22}O$, was determined by mass spectrometry ($[M]^+$ obs. 266.1673 (71%) req. 266.1665). The IR spectrum of 3 revealed the presence of an aromatic ring system (1574 , 815 and 784 cm^{-1}) and

strong absorptions for an ether linkage (1260 and 1033 cm^{-1}). The UV spectrum showed the presence of a substituted naphthalene ($\lambda_{\text{max}}^{\text{EtOH}}$ 241, 306, 322 and 336 nm). The ^1H NMR spectrum, in CDCl_3 , showed one singlet for a geminal dimethyl group at δ 1.36 (6H); an aromatic methyl substituent at δ 2.38 (3H, s); three aromatic protons: two *ortho* with respect to each other at δ 7.26 (1H, *d*, $J = 8.5\text{ Hz}$) and δ 7.70 (1H, *d*, $J = 8.5\text{ Hz}$) and a third one, a singlet at δ 6.92 (1H) for the hydrogen *ortho* to the alkoxy group. Two triplets at δ 3.10 (2H, $J = 5\text{ Hz}$) and δ 4.36 (2H, $J = 5\text{ Hz}$) were assigned to an $\text{ArCH}_2\text{CH}_2\text{O}$ -moiety suggesting that the oxygen of 3 belongs to a dihydropyran ring fused to the aromatic nucleus. These data and biogenetic considerations are consistent with structure 3 for the naphthalenic nor-cleistanthane diterpene.

The Wolff-Kishner reduction of 1 afforded a product that was identical in all aspects with natural naphthalenic norditerpene 3.



EXPERIMENTAL

Mps are uncorr. UV spectra were recorded in 95% EtOH IR spectra in KBr discs. ^1H and ^{13}C NMR spectra were recorded at 100 and 25.2 MHz, respectively, and chemical shifts (δ -values, ppm) measured from TMS as int. standard.

Isolation of 3. Chromatography of the hexane extract of the trunk, roots and leaf-sheaths of *V. epidendroides* and *V. phalocarpa* collected in Diamantina, Minas Gerais, Brazil, yielded the naphthalene norcleistanthane 3, mp 136 – 138° ; $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 1574, 1450, 1388, 1366, 1318, 1272, 1231, 1163, 1111, 1032, 934, 892, 847, 815 and 784. UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm: 241, 306, 322 and 336. ^1H NMR (100

MHz, CDCl_3 : δ 1.36 (6H, s), 1.55 at 2.06 (4H, m), 2.38 (3H, s), 2.98 (2H, t, $J = 6$ Hz), 3.10 (2H, t, $J = 5$ Hz), 4.36 (2H, t, $J = 5$ Hz), 6.92 (1H, s), 7.26 (1H, d, $J = 8.5$ Hz) and 7.70 (1H, d, $J = 8.5$ Hz). ^{13}C NMR (25.2 MHz, CDCl_3): δ 18.3 (q), 19.5 (t), 26.2 (t), 26.6 (t), 31.4 (q), 31.4 (q), 34.1 (s), 38.8 (t), 65.5 (t), 108.3 (d), 120.2 (s), 121.2 (d), 123.0 (s), 127.3 (s), 128.6 (d), 128.6 (s), 131.3 (s), 142.4 (s) and 150.5 (s). MS m/z (rel. int.): 266 $[\text{M}]^+$ (71), 251 (100), 236 (14), 221 (16), 210 (9) and 207 (8).

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ISOCARNOSOL, A DITERPENE FROM *SALVIA LANIGERA*

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Key Word Index—*Salvia lanigera*; Labiatae; diterpene; isocarnosol.

Abstract—Isocarnosol, a novel diterpene from the leaves of *Salvia lanigera*, has been characterized.

INTRODUCTION

The leaves of *Salvia lanigera* Poir are used as an aromatic tea for a variety of abdominal troubles. A number of diterpenes and diterpene-quinones have been isolated from many *Salvia* species [1–9]. Four diterpene-quinones of the royleanone type have been isolated [10] from *S. lanigera*.

RESULTS AND DISCUSSION

Column chromatography (silica gel) of the petrol extract of the leaves of *S. lanigera* gave a number of fractions. Isocarnosol crystallized out from the chloroform fraction and was recrystallized from methanol to give white crystals, mp 216°. Isocarnosol was given the structure 1 primarily on the basis of ^1H NMR data.

Isocarnosol (1) $\text{C}_{20}\text{H}_{26}\text{O}_4$ has the same molecular formula as carnosol (2) which has been isolated from *Salvia triloba* [11], but the physical properties of the two compounds are different. The IR spectrum of isocarnosol showed a striking resemblance in many of its features to that of carnosol, which exhibits the presence of a 6-membered lactone ring (1730 and 1205 cm^{-1}), an aromatic ring (3050, 1600 and 1500 cm^{-1}) and phenolic hydroxyl groups at 3350 cm^{-1} .

The ^1H NMR spectrum of 1 in CDCl_3 containing a small amount of DMSO- d_6 indicated the presence of an isopropyl group (δ 1.18, $J = 6.7$ Hz), two C-methyl (0.85 and 0.89) and a characteristic low field aromatic C-H at 6.63. The aromatic C-H shifted to low field at δ 7.0 in the diacetate derivative indicating its proximity to one of the phenolic hydroxyl groups. The phloroglucinol test for an *o*-diphenol function [12] was negative for isocarnosol,

thus leaving the other two isomers (*meta* and *para*) as possibilities. These conclusions were supported by the interpretation of the ^{13}C NMR spectrum of isocarnosol in which the unsubstituted aromatic carbon resonates at δ 111.78. The up-field shift (δ 111.78) for the unsubstituted carbon atom is characteristic of an aromatic carbon adjacent to an oxygen substituent. C-14 in carnosol (2) should resonate in the same region as does the same carbon in the diacetate derivative of aethiopinone (3) [13] or in carnosolone (4) [14] (δ 124.0 and 119.7 respectively).

At this stage the compound was identified as isocarnosol (1) or its *meta*-isomer. A structure of the sempervirol type was excluded on biosynthetic grounds as none of the more than 40 diterpene and diterpene-quinones isolated from various *Salvia* species has this structure [15]. However, examination of the IR and NMR data of the oxidation product of 1 showed that this compound was of the *p*-quinonoid type. Thus the IR spectrum of the major product of the bromine oxidation of 1 showed a band at 1660 cm^{-1} (shoulder) typical of the quinonoid structure and an intense band at 1760 cm^{-1} attributable to carbonyl lactone absorption. The appearance of a doublet signal at δ 6.67 ($J \sim 1.2$ Hz) in the ^1H NMR spectrum indicates the structure 5.

The small coupling constant for proton H-12 (~ 1.2 Hz) is due to allylic coupling with a hydrogen at C-15, and this situation could not be expected in the *m*-quinonoid type 6 (i.e. H-11 should resonate as a singlet). The ^{13}C NMR spectrum of isocarnosol (Table 1) indicates that one of the C-4 methyl groups is in a 1,3-diaxial relationship with the carboxyl lactone at C-10 and this can happen only if the A/B rings are *trans*-fused [16]. That compounds 1 and 2 have the same absolute configuration